

# Multivariate Statistical Analysis of Heavy Metals and other Hydro Chemical Characteristics in Industrially Polluted Groundwater Resources of Mettur, India

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**Abstract:** Mettur an industrial town is located in Salem district of Tamil Nadu, India. The major industries, namely; Chemplast, The Madras Aluminum Company and Mettur Thermal Power Plant are located in Mettur town. In addition, several chemical industries are situated on the banks of the River Cauvery as part of Small Industries Development Corporation (SIDCO) industrial estate. With this background, the present study attempts to investigate the usefulness of multivariate statistical analysis for the assessment of complex data sets to get better information about an influence of various parameters on groundwater chemistry and heavy metal concentrations. Groundwater samples were collected during pre-monsoon (June 2013) and post-monsoon (February 2014) seasons and analysed for heavy metals and other hydro chemical characteristics. The results of the multivariate statistical analysis show that the sampling site have different characteristics in terms of heavy metals studied and indicate that each region receives pollution from different sources. The high level of some of the heavy metals and hydrochemical recorded in certain pollution packets ultimately implies an alarming situation in the viewpoint of health risk. The notable differences in the quality of groundwater between the sampling sites indicate possible land use influences.

**Keywords:** Groundwater, Pollution, Multivariate statistics, Industrial pollution, Agrochemicals

## Introduction

Groundwater resources are highly crucial to a range of human needs such as domestic, agricultural and industrial purposes. The consumption rate of groundwater is increasing gradually in the areas where surface water sources are not enough to meet the demands<sup>1</sup>. According to recent reports, several states in India are dependent on groundwater to meet out their 90% of domestic and agricultural needs<sup>2-3</sup>. Dispose of untreated/partially treated industrial effluents, urban wastes and agrochemicals (fertilizers, herbicides and pesticides) are the primary causes of groundwater contamination<sup>4</sup>. Although the industrial sector only accounts for three per cent of the annual water withdrawals in India, it causes severe water pollution in most of the freshwater and groundwater resources. Among the pollutants, heavy

metals are ubiquitous in nature and gain scientific interest due to its unique characteristics such as high reactivity, lithophilic nature, toxicity and non-biodegradability<sup>5</sup>. The term “heavy metal” refers to any metallic chemical element that has a relatively high density and is toxic at low concentrations.

Decline of water quality in general and groundwater in particular is of great concern in India. Studies on groundwater quality monitoring would help to identify safe zones for drinking water and provide solution to the quality problems in groundwater by means of hydrogeological and geochemical data. Although statistical approaches are recommended for its usefulness in the monitoring of groundwater resources and its related pollution studies have not been utilized much. Multivariate statistical approach allows deriving simple relations which are much useful in characterizing and getting first hand information of the groundwater system than actually going through complex methodologies.

## Experimental

The study area Mettur is one among the important industrial centres of Tamil Nadu. It is located at 11° 45' of the Northern Latitudes and 77° 45' of the Eastern Longitude (Figure 1). It has its uniqueness in the distribution of a variety of physiographic features ranging from extensive hilly areas to undulating plains. Three largest industries namely Chemplast, Madras Aluminum Company and the Mettur Thermal Power Plant are located in Mettur town. In addition, numerous chemical industries are located on the banks of the River Cauvery in Mettur as a part of Small Scale Industries Development Corporation (SIDCO) Industrial Estate. In this background, the present study attempts to investigate the heavy metals and other hydro chemical characteristics in groundwater samples in the Mettur industrial region using multivariate data analysis.

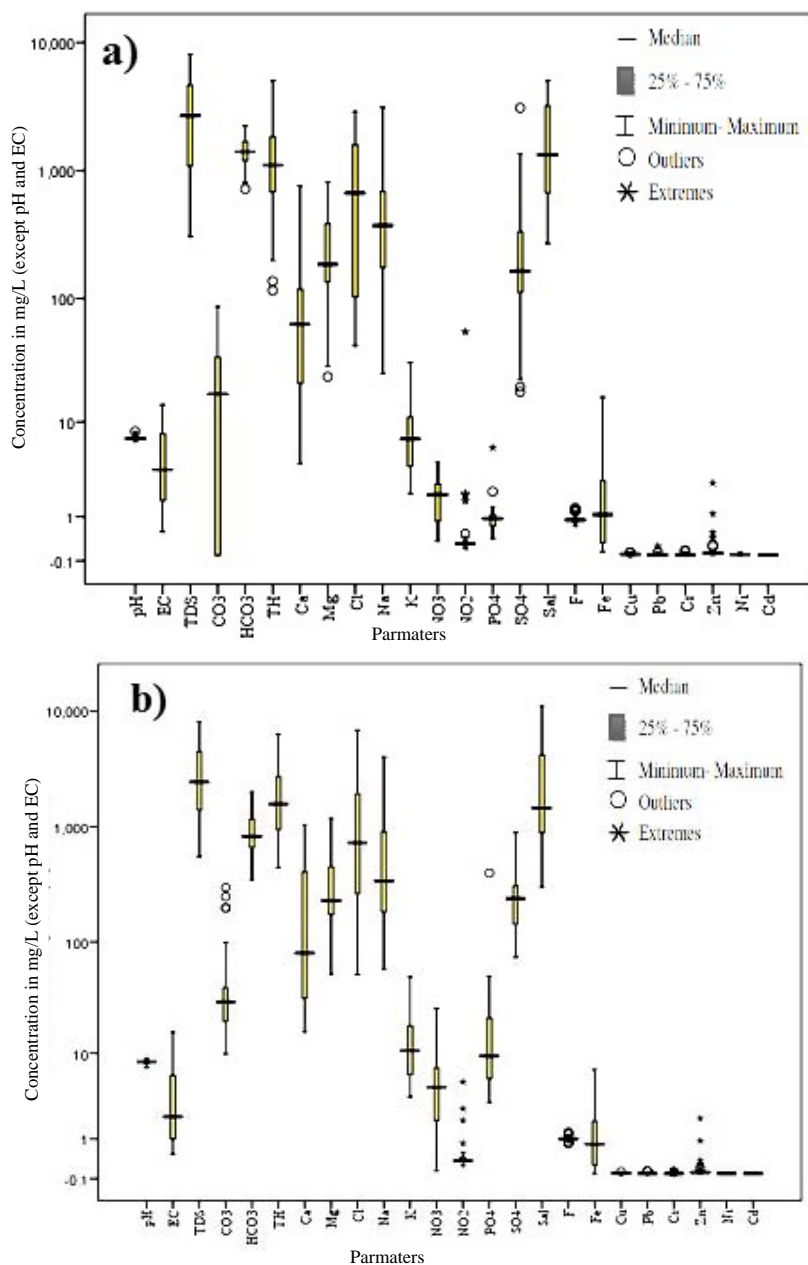


**Figure 1.** Sampling location and map of the Mettur industrial area

### *Sampling and Methodology*

Groundwater samples were collected in acid washed polythene bottles from 49 stations during pre-monsoon (June 2013) and post-monsoon season (February 2014) in Mettur

industrial area. The sampling locations are shown in Figure 1. Water samples were analyzed for pH, Electrical Conductivity (EC), Total Dissolved Solids (TDS), total hardness, chloride (Cl), alkalinity, calcium (Ca), magnesium (Mg), sodium (Na), potassium (K), nitrite ( $\text{NO}_2$ ) nitrates ( $\text{NO}_3$ ), sulfate ( $\text{SO}_4$ ), phosphate ( $\text{PO}_4$ ) and fluoride (F) as per standard methods<sup>6</sup>.



**Figure 2.** Hydrochemical characteristics of groundwater during; a) pre-monsoon and b) post-monsoon season

For the heavy metals analysis, water samples (50 mL) were subjected to concentrated  $\text{HNO}_3$  digestion until the solution became transparent and atomic absorption spectrophotometrically analyzed (Shimadzu AA6300). The results obtained were subjected to statistical analysis, such as paired *T-test*, cluster analysis and factor analysis using Statistical Package for Social Scientists (SPSS) 16<sup>th</sup> version.

## Results and Discussion

### *Physical properties of groundwater samples*

Groundwater qualities of forty-nine samples representing Mettur industrial area during pre-monsoon and post-monsoon seasons are given in Figure 2. The pH of the groundwater samples were observed as slightly acidic to alkaline nature (6.67-8.90). Electrical conductivity (EC) level of the samples ranged between 0.48 and 15.7 mS/cm. Mean level of total dissolved solids (TDS) was figured as 2896 and 2977 mg/L in pre-monsoon and post-monsoon season, respectively. Notably higher level of EC and TDS recorded in post-monsoon season samples implies monsoonal influenced leaching of industrial effluents and agrochemicals in the study area.

### *Major ions concentration*

Concentration of calcium (Ca), magnesium (Mg), sodium (Na) and potassium (K) were figured in the range of 4.2-1044 mg/L, 25.3-1166.7 mg/L, 25.3-4004 mg/L and 2.02-49.4 mg/L, respectively. Mg content was considerably higher than other cations (Ca and K) was probably due to the leaching of effluents from nearby agrochemical industry. Concentration of cation was observed in the order of  $\text{Na} > \text{Mg} > \text{Ca} > \text{K}$ , whereas anion level was noted as  $\text{HCO}_3 > \text{Cl} > \text{SO}_4 > \text{CO}_3 > \text{NO}_3 > \text{NO}_2 > \text{PO}_4$  and  $\text{Cl} > \text{HCO}_3 > \text{SO}_4 > \text{CO}_3 > \text{PO}_4 > \text{NO}_3 > \text{NO}_2$  in pre-monsoon and post-monsoon, respectively. During post-monsoon season, total hardness content of all water samples exceeded the permissible limit of 300 mg/L.

Mean level of chloride was figured as 852 and 1188 mg/L in pre-monsoon and post-monsoon season, respectively. In general, notably very high concentration of Cl recorded in few locations (G19, G21 and G23) probably contributed by leaching of industrial effluents in the vicinity, as these sample locations are very close to a cluster of industries such as Chemplast, Malco and *etc.*, indicates suspected groundwater contamination due to industrial effluents. Supportively, the study area is underlined by silicate minerals which are known for a lesser amount of chloride thus natural enrichment of chloride in this region is trace or negligible<sup>7</sup>.

Phosphate ( $\text{PO}_4$ ) content in 70% of sampling sites in pre-monsoon and all the sampling sites in post-monsoon exceeded the permissible limit (0.1 mg/L) of US Public Health Standards<sup>8</sup>. Considerably higher concentration of  $\text{PO}_4$  was recorded during post-monsoon shows notable high  $\text{PO}_4$  input from manmade sources such as discharge of untreated or partially treated industrial effluents and municipal sewage in addition to agrochemical leaching. The level of Sulphate ( $\text{SO}_4$ ) falls in the range from 17.8 to 3090 mg/L and 73.2 to 893.2 mg/L in pre-monsoon and post-monsoon, respectively. Fluoride content in all samples falls (0.7-1.35 mg/L) within the optimum concentration of 1.5 mg/L, as recommended by WHO. Correlation analysis shows strong positive relationship between EC and TDS with Na, Ca, Mg,  $\text{HCO}_3$ , Cl and total hardness implies these elements contributes in ionic load of the groundwater (Table 1).

**Table 1.** Correlation matrix for hydro-chemical characteristics

|                        | pH            | EC            | TDS           | CO <sub>3</sub> | HCO <sub>3</sub> | TH            | Ca            | Mg            | Cl            | Na     | K      | NO <sub>3</sub> | NO <sub>2</sub> | PO <sub>4</sub> | SO <sub>4</sub> | Sal  | F |
|------------------------|---------------|---------------|---------------|-----------------|------------------|---------------|---------------|---------------|---------------|--------|--------|-----------------|-----------------|-----------------|-----------------|------|---|
| <b>a) Pre-Monsoon</b>  |               |               | mg/L          | mg/L            | mg/L             | mg/L          | mg/L          | mg/L          | mg/L          | mg/L   | mg/L   | mg/L            | mg/L            | mg/L            | mg/L            | mg/L |   |
| pH                     | 1             |               |               |                 |                  |               |               |               |               |        |        |                 |                 |                 |                 |      |   |
| EC                     | -0.44         | 1             |               |                 |                  |               |               |               |               |        |        |                 |                 |                 |                 |      |   |
| TDS                    | -0.45         | 0.96**        | 1             |                 |                  |               |               |               |               |        |        |                 |                 |                 |                 |      |   |
| CO <sub>3</sub>        | <b>0.90**</b> | -0.42         | -0.43         | 1               |                  |               |               |               |               |        |        |                 |                 |                 |                 |      |   |
| HCO <sub>3</sub>       | -0.41         | 0.53**        | 0.49**        | -0.29           | 1                |               |               |               |               |        |        |                 |                 |                 |                 |      |   |
| TH                     | -0.42         | <b>0.77**</b> | <b>0.81**</b> | -0.33           | 0.48**           | 1             |               |               |               |        |        |                 |                 |                 |                 |      |   |
| Ca                     | -0.37         | <b>0.76**</b> | <b>0.75**</b> | -0.32           | 0.23             | <b>0.68**</b> | 1             |               |               |        |        |                 |                 |                 |                 |      |   |
| Mg                     | -0.36         | 0.63**        | <b>0.68**</b> | -0.27           | 0.50**           | <b>0.94**</b> | 0.41**        | 1             |               |        |        |                 |                 |                 |                 |      |   |
| Cl                     | -0.46         | <b>0.96**</b> | <b>0.93**</b> | -0.43           | 0.52**           | <b>0.77**</b> | 0.77**        | 0.62**        | 1             |        |        |                 |                 |                 |                 |      |   |
| Na                     | -0.26         | 0.60**        | 0.59**        | -0.33           | 0.24             | 0.33*         | 0.53*         | 0.18          | 0.60**        | 1      |        |                 |                 |                 |                 |      |   |
| K                      | -0.16         | 0.26          | 0.24          | -0.14           | 0.18             | 0.29*         | 0.19          | 0.27          | 0.18          | 0.03   | 1      |                 |                 |                 |                 |      |   |
| NO <sub>3</sub>        | -0.05         | 0.40**        | 0.41**        | -0.01           | 0.26             | 0.46**        | 0.26          | 0.46**        | 0.30*         | 0.20   | 0.24   | 1               |                 |                 |                 |      |   |
| NO <sub>2</sub>        | -0.12         | 0.28          | 0.26          | -0.12           | 0.13             | 0.02          | 0.27          | -0.08         | 0.313*        | 0.12   | 0.14   | -0.02           | 1               |                 |                 |      |   |
| PO <sub>4</sub>        | 0.09          | -0.07         | -0.11         | 0.16            | -0.02            | -0.09         | -0.1          | -0.07         | -0.1          | -0.09  | -0.08  | -0.02           | -0.01           | 1               |                 |      |   |
| SO <sub>4</sub>        | -0.18         | 0.49**        | 0.45**        | -0.19           | 0.39**           | 0.42**        | 0.21          | 0.43**        | 0.41**        | 0.31*  | 0.41** | 0.42**          | 0.25            | -0.15           | 1               |      |   |
| Sal                    | -0.41         | <b>0.98**</b> | <b>0.94**</b> | -0.38           | 0.53**           | <b>0.76**</b> | <b>0.73**</b> | <b>0.63**</b> | <b>0.96**</b> | 0.59** | 0.27   | 0.40**          | 0.291*          | -0.06           | 0.50**          | 1    |   |
| F                      | 0.14          | 0.12          | 0.06          | 0.15            | 0.14             | -0.06         | 0.01          | -0.08         | 0.05          | 0.25   | -0.04  | -0.03           | 0.09            | -0.13           | 0.40**          | 0.11 | 1 |
| <b>b) Post-Monsoon</b> |               |               |               |                 |                  |               |               |               |               |        |        |                 |                 |                 |                 |      |   |
| pH                     | 1             |               |               |                 |                  |               |               |               |               |        |        |                 |                 |                 |                 |      |   |
| EC                     | -0.3          | 1             |               |                 |                  |               |               |               |               |        |        |                 |                 |                 |                 |      |   |
| TDS                    | -0.17         | 0.74**        | 1             |                 |                  |               |               |               |               |        |        |                 |                 |                 |                 |      |   |
| CO <sub>3</sub>        | -0.14         | 0.53**        | 0.65**        | 1               |                  |               |               |               |               |        |        |                 |                 |                 |                 |      |   |
| HCO <sub>3</sub>       | -0.20         | 0.34*         | 0.38**        | 0.64**          | 1                |               |               |               |               |        |        |                 |                 |                 |                 |      |   |
| TH                     | -0.25         | 0.66**        | <b>0.89**</b> | 0.54**          | 0.34*            | 1             |               |               |               |        |        |                 |                 |                 |                 |      |   |
| Ca                     | -0.31         | <b>0.85**</b> | <b>0.77**</b> | 0.50**          | 0.20             | 0.71**        | 1             |               |               |        |        |                 |                 |                 |                 |      |   |
| Mg                     | -0.12         | 0.30*         | <b>0.68**</b> | 0.39**          | 0.33*            | <b>0.86**</b> | 0.28          | 1             |               |        |        |                 |                 |                 |                 |      |   |
| Cl                     | -0.27         | <b>0.86**</b> | <b>0.73**</b> | 0.48**          | 0.27             | 0.64**        | 0.81**        | 0.30*         | 1             |        |        |                 |                 |                 |                 |      |   |
| Na                     | -0.05         | 0.65**        | <b>0.81**</b> | 0.69**          | 0.40*            | 0.56**        | 0.62**        | 0.32*         | 0.73**        | 1      |        |                 |                 |                 |                 |      |   |
| K                      | -0.15         | 0.40**        | 0.41**        | 0.34*           | 0.07             | 0.42**        | 0.42**        | 0.27          | 0.29*         | 0.31*  | 1      |                 |                 |                 |                 |      |   |
| NO <sub>3</sub>        | 0.20          | 0.04          | -0.10         | -0.13           | -0.08            | -0.08         | -0.06         | -0.07         | 0.06          | -0.14  | -0.08  | 1               |                 |                 |                 |      |   |
| NO <sub>2</sub>        | -0.04         | 0.43**        | 0.42**        | 0.38**          | 0.34*            | 0.47**        | 0.48**        | 0.30*         | 0.59**        | 0.50** | 0.07   | -0.06           | 1               |                 |                 |      |   |
| PO <sub>4</sub>        | -0.03         | 0.08          | 0.08          | 0.33*           | 0.32*            | 0.05          | 0.01          | 0.06          | -0.11         | -0.03  | 0.06   | 0.03            | -0.07           | 1               |                 |      |   |
| SO <sub>4</sub>        | -0.20         | 0.48**        | 0.59**        | 0.18            | 0.13             | 0.48**        | 0.45**        | 0.34*         | 0.53**        | 0.60** | 0.34*  | -0.03           | 0.49**          | -0.14           | 1               |      |   |
| Sal                    | -0.21         | 0.88**        | 0.76**        | 0.42**          | 0.28*            | 0.66**        | 0.81**        | 0.33*         | 0.88**        | 0.75** | 0.44** | -0.03           | 0.53**          | -0.16           | .64**           | 1    |   |
| F                      | -0.06         | 0.07          | 0.02          | 0.08            | 0.20             | 0.02          | -0.02         | 0.04          | -0.01         | 0.04   | 0.08   | -0.08           | 0.06            | 0.09            | 0.11            | 0.02 | 1 |

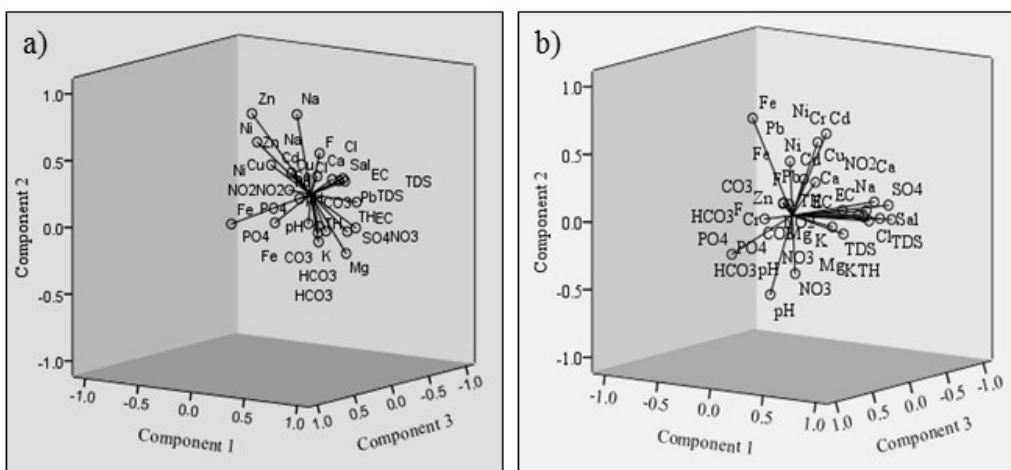
\*\* Correlation is significant at the 0.01 level (2-tailed), \* Correlation is significant at the 0.05 level (2-tailed)

### Trace elements concentration

Maximum concentration of Fe, Cu, Cr, Zn, Pb, Ni and Cd were measured as 13.2 mg/L, 0.05 mg/L, 0.1 mg/L, 2.68 mg/L, 0.18 mg/L, 0.04 mg/L and 0.008 mg/L, respectively. Mean level of the heavy metals in water samples were observed as follows, Pre - monsoon: Fe > Zn > Cu > Cr > Pb > Ni > Cd and Post-monsoon: Fe > Zn > Cu > Pb > Cr > Ni > Cd. Paired T-test yielded significant difference in heavy metals concentration between the seasons studied (Pre-monsoon and post-monsoon). In addition to industrial pollution, Zn has agricultural origin coming from extensive use of pesticides and fertilizers by the farmers to improve crop yields in the area<sup>9</sup>.

### Principle component analysis

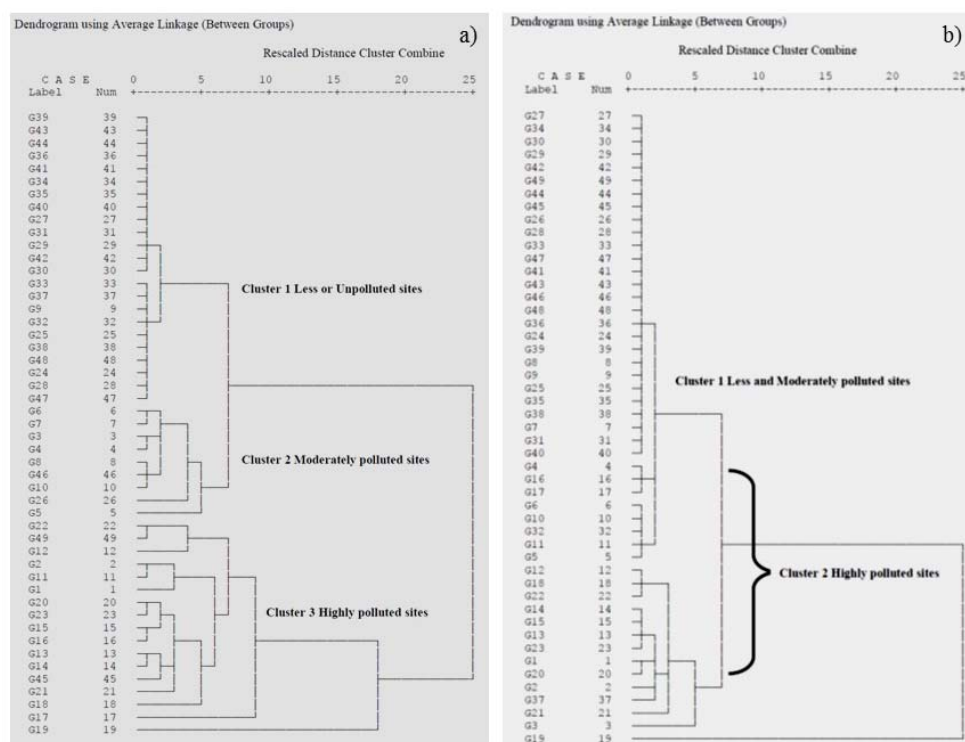
Factor analysis is a multivariate statistical and it attempts to explain the variance of a large set of interrelated variables with smaller set of independent variables<sup>10</sup>. The factor analysis for pre-monsoon data has extracted four factors which explained 61.34% of the variance in the data set (Figure 3). Factor 1 accounted for 27.46% of representation with high positive loadings of EC, TDS, TH, Ca, Mg, Cl, NO<sub>3</sub>, Salinity and SO<sub>4</sub>. This factor loading suggests diverse role of weathering and anthropogenic influences on these parameters<sup>11</sup>. Factor 2 accounted for 14.39% of the total variance with loadings of Na, Zn and Ni. Factor 3 explains 10.26% of total variance with negative loading of pH and CO<sub>3</sub>. Factor 4 yielded 9.22% of total variance with higher loadings of PO<sub>4</sub>, Fe, Pb and Cr. Agricultural discharge containing residual of pesticides and fertilizers also contains metals also is seen as another important source of nutrients and heavy metal enrichment in groundwater. For example, Zn, Cu and Cd level may also relate to fungicide, pesticides and herbicide applications, which usually contain high levels of Zn salts and Cu arsenates<sup>12-14</sup>. Notably, some phosphate fertilizers contain potentially toxic elements, including As, Cd, Cr, Pd, Hg, Ni, and V and some pesticides have contained Cu and As as part of their formulation<sup>15</sup>. For Cr, inorganic fertilizers were the largest single source followed by sewage sludge.



**Figure 3.** Factor analysis for a) pre-monsoon and b) post-monsoon season

The factor analysis for Post-monsoon data has explained 67.04% of the total variance with five factors (Figure 3). Factor 1 accounted for 30.12% of representation with high loadings of EC, TDS, total hardness, Ca, Mg, Cl, NO<sub>2</sub>, SO<sub>4</sub> and salinity. This factor explains

complex nature of hydro geochemical processes and its influences factors (natural and anthropogenic) on these parameters. Since, the study area is not known for geogenic sources of Cl, Ca, Mg and SO<sub>4</sub> probably these elements may enriched by anthropogenic activities such as industrial effluent disposal and agrochemical application<sup>7,16</sup>. Since, the industrial effluents are known for higher concentration of Ca, TDS, Mg, Cl and SO<sub>4</sub> Factor 2 accounted for 10.44% of the total variance with loadings of Fe, Ni and Cd. Factor 3 explains 9.79% of total variance with loadings of CO<sub>3</sub>, HCO<sub>3</sub>, PO<sub>4</sub> and Cr. Factor 4 and 5 contains loadings Cu and Zn. It reveals notable contributions from agrochemicals runoff. It is worthy to mention that according to recent report the excessive utilization of fertilisers and pesticides for agricultural activities in the study area has resulted in a localised enrichment of NO<sub>3</sub>, PO<sub>4</sub>, SO<sub>4</sub>, F and heavy metals in the groundwater resources<sup>16</sup>.



**Figure 4.** Cluster analysis for a) pre-monsoon and b) post-monsoon season

### Cluster analysis

Cluster analysis (CA) was computed to identify the similarity groups between the sampling sites. In pre-monsoon season (Figure 4a), Cluster 1 grouped less and unpolluted sampling sites. Cluster 2 related to moderately polluted sites whereas cluster 3 formed by the highly contaminated sites. Interestingly, post-monsoon season cluster analysis yielded two clusters namely; cluster 1 contains less and moderately polluted sampling sites and cluster 2 is associated with highly contaminated sites (Figure 4b). It implies monsoonal influenced point and non-point source pollutant leaching process and groundwater flow ultimately expanded the polluted zone in post-monsoon season. The notable differences in the quality of groundwater between the sampling sites indicate possible land use influences.

## Conclusion

The groundwater quality of the Mettur industrial area is highly impaired as a result of unregulated manmade activities. Taking into account of the high concentrations of TDS, Cl, Ca, Mg, Na, PO<sub>4</sub> and SO<sub>4</sub>, the present situation imply serious risks to human health. Factor analysis suggested the disposal of industrial effluents; municipal sewage and agrochemical leaching are the major influencing factors of the ground water quality in the study area. In Cluster analysis, the notable differences in the quality of groundwater between the sampling sites indicate possible land use influences on water quality. The present study also stresses the importance of the periodical monitoring of the groundwater quality to assess the extent of pollution and other inhibitory chemicals which affect the groundwater resources in Mettur industrial region.

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